## **Environmental Protection Agency**

 $C_{\text{corr}} = C_{\text{meas}} \times (21 - X/21 - Y)$ where:

 $C_{\text{corr}}$ =the concentration corrected for oxygen.

 $C_{\mathrm{meas}}$ =the concentration uncorrected for oxvgen.

X=the volumetric oxygen concentration in percentage to be corrected to (8 percent for recovery furnaces and 10 percent for lime kilns, incinerators, or other devices).

Y=the measured 12-hour average volumetric oxygen concentration.

- (4) Record once per shift measurements obtained from the continuous monitoring devices installed under paragraph (b)(2) of this section.
- (d) For the purpose of reports required under §60.7(c), any owner or operator subject to the provisions of this subpart shall report semiannually periods of excess emissions as follows:
- (1) For emissions from any recovery furnace periods of excess emissions are:
- (i) All 12-hour averages of TRS concentrations above 5 ppm by volume for straight kraft recovery furnaces and above 25 ppm by volume for cross recovery furnaces.
- (ii) All 6-minute average opacities that exceed 35 percent.
- (2) For emissions from any lime kiln, periods of excess emissions are all 12-hour average TRS concentration above 8 ppm by volume.
- (3) For emissions from any digester system, brown stock washer system, multiple-effect evaporator system, or condensate stripper system periods of excess emissions are:
- (i) All 12-hour average TRS concentrations above 5 ppm by volume unless the provisions of §60.283(a)(1) (i), (ii), or (iv) apply; or
- (ii) All periods in excess of 5 minutes and their duration during which the combustion temperature at the point of incineration is less than 650 °C (1200 °F), where the provisions of §60.283(a)(1)(iii) apply.
- (e) The Administrator will not consider periods of excess emissions reported under paragraph (d) of this section to be indicative of a violation of §60.11(d) provided that:
- (1) The percent of the total number of possible contiguous periods of excess emissions in a quarter (excluding periods of startup, shutdown, or malfunction and periods when the facility is

not operating) during which excess emissions occur does not exceed:

- (i) One percent for TRS emissions from recovery furnaces.
- (ii) Six percent for average opacities from recovery furnaces.
- (2) The Administrator determines that the affected facility, including air pollution control equipment, is maintained and operated in a manner which is consistent with good air pollution control practice for minimizing emissions during periods of excess emissions.
- (f) The procedures under §60.13 shall be followed for installation, evaluation, and operation of the continuous monitoring systems required under this section.
- (1) All continuous monitoring systems shall be operated in accordance with the applicable procedures under Performance Specifications 1, 3, and 5 of appendix B to this part.
- (2) Quarterly accuracy determinations and daily calibration drift tests shall be performed in accordance with Procedure 1 of appendix F to this part.

[43 FR 7572, Feb. 23, 1978, as amended at 51 FR 18545, May 20, 1986; 65 FR 61759, Oct. 17, 20001

### § 60.285 Test methods and procedures.

- (a) In conducting the performance tests required in §60.8, the owner or operator shall use as reference methods and procedures the test methods in appendix A of this part or other methods and procedures in this section, except as provided in §60.8(b). Acceptable alternative methods and procedures are given in paragraph (f) of this section.
- (b) The owner or operator shall determine compliance with the particulate matter standards in §60.282(a) (1) and (3) as follows:
- (1) Method 5 shall be used to determine the particulate matter concentration. The sampling time and sample volume for each run shall be at least 60 minutes and 0.90 dscm (31.8 dscf). Water shall be used as the cleanup solvent instead of acetone in the sample recovery procedure. The particulate concentration shall be corrected to the appropriate oxygen concentration according to §60.284(c)(3).
- (2) The emission rate correction factor, integrated sampling and analysis

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procedure of Method 3B shall be used to determine the oxygen concentration. The gas sample shall be taken at the same time and at the same traverse points as the particulate sample.

- (3) Method 9 and the procedures in §60.11 shall be used to determine opacity.
- (c) The owner or operator shall determine compliance with the particular matter standard in §60.282(a)(2) as follows:
- (1) The emission rate (E) of particulate matter shall be computed for each run using the following equation:

 $E{=}c_s\,Q_{sd}{/}BLS$ 

where:

E=emission rate of particulate matter, g/kg (lb/ton) of BLS.

 $c_s$  = Concentration of particulate matter, g/dscm (lb/dscf).

 $Q_{sd}$ =volumetric flow rate of effluent gas, dscm/hr (dscf/hr).

BLS=black liquor solids (dry weight) feed rate, kg/hr (ton/hr).

- (2) Method 5 shall be used to determine the particulate matter concentration  $(c_s)$  and the volumetric flow rate  $(Q_{sd})$  of the effluent gas. The sampling time and sample volume shall be at least 60 minutes and 0.90 dscm (31.8 dscf). Water shall be used instead of acetone in the sample recovery.
- (3) Process data shall be used to determine the black liquor solids (BLS) feed rate on a dry weight basis.
- (d) The owner or operator shall determine compliance with the TRS standards in §60.283, except §60.283(a)(1)(vi) and (4), as follows:
- (1) Method 16 shall be used to determine the TRS concentration. The TRS concentration shall be corrected to the appropriate oxygen concentration using the procedure in §60.284(c)(3). The sampling time shall be at least 3 hours, but no longer than 6 hours.
- (2) The emission rate correction factor, integrated sampling and analysis procedure of Method 3B shall be used to determine the oxygen concentration. The sample shall be taken over the same time period as the TRS samples.
- (3) When determining whether a furnace is a straight kraft recovery furnace, or a cross recovery furnace, TAPPI Method T.624 (incorporated by reference—see §60.17) shall be used to determine sodium sulfide, sodium hydroxide, and sodium carbonate. These

determinations shall be made 3 times daily from the green liquor, and the daily average values shall be converted to sodium oxide ( $Na_20$ ) and substituted into the following equation to determine the green liquor sulfidity:

GLS = 
$$100 C_{Na_2S} / (C_{Na_2S} C_{NaOH} C_{Na_2CO_3})$$

Where

GLS=green liquor sulfidity, percent.

 $C_{Na_2S} {=} concentration$  of Na\_S as Na\_O, mg/ liter (gr/gal).

 $C_{NaOH}$ =concentration of NaOH as Na<sub>2</sub>O, mg/liter (gr/gal).

 $C_{Na2CO3}$ =concentration of  $Na_2CO_3$  as  $Na_2O$ , mg/liter (gr/gal).

- (e) The owner or operator shall determine compliance with the TRS standards in  $\S 60.283(a)(1)(vi)$  and (4) as follows:
- (1) The emission rate (E) of TRS shall be computed for each run using the following equation:

 $E = C_{TRS} \ F \ Q_{sd} / P$ 

where:

E=emission rate of TRS, g/kg (lb/ton) of BLS or ADP.

 $C_{TRS}$ =average combined concentration of TRS, ppm.

F = conversion factor, 0.001417 g  $H_2S/m^3$ -ppm (8.846  $\times$  10<sup>-8</sup> lb  $H_2S/ft^3$ -ppm).

Q<sub>sd</sub>=volumetric flow rate of stack gas, dscm/hr (dscf/hr).

P=black liquor solids feed or pulp production rate, kg/hr (ton/hr).

- (2) Method 16 shall be used to determine the TRS concentration ( $C_{TRS}$ ).
- (3) Method 2 shall be used to determine the volumetric flow rate  $(Q_{sd})$  of the effluent gas.
- (4) Process data shall be used to determine the black liquor feed rate or the pulp production rate (P).
- (f) The owner or operator may use the following as alternatives to the reference methods and procedures specified in this section:
- (1) For Method 5, Method 17 may be used if a constant value of 0.009 g/dscm (0.004 gr/dscf) is added to the results of Method 17 and the stack temperature is no greater than 204 °C (400 °F).
- (2) In place of Method 16, Method 16A or 16B may be used.

[54 FR 6673, Feb. 14, 1989; 54 FR 21344, May 17, 1989, as amended at 55 FR 5212, Feb. 14, 1990; 65 FR 61759, Oct. 17, 2000]

## Subpart CC—Standards of Performance for Glass Manufacturing Plants

# § 60.290 Applicability and designation of affected facility.

- (a) Each glass melting furnace is an affected facility to which the provisions of this subpart apply.
- (b) Any facility under paragraph (a) of this section that commences construction or modification after June 15, 1979, is subject to the requirements of this subpart.
- (c) This subpart does not apply to hand glass melting furnaces, glass melting furnaces designed to produce less than 4.55 Mg (5 tons) of glass per day and all-electric melters.

 $[45~\mathrm{FR}~66751,~\mathrm{Oct.}~7,~1980,~\mathrm{as}$  amended at  $65~\mathrm{FR}~61759,~\mathrm{Oct.}~17,~2000]$ 

#### § 60.291 Definitions.

As used in this subpart, all terms not defined herein shall have the meaning given them in the Act and in subpart A of this part, unless otherwise required by the context.

All-electric melter means a glass melting furnace in which all the heat required for melting is provided by electric current from electrodes submerged in the molten glass, although some fossil fuel may be charged to the furnace as raw material only.

Borosilicate recipe means glass product composition of the following approximate ranges of weight proportions: 60 to 80 percent silicon dioxide, 4 to 10 percent total  $R_2O$  (e.g.,  $Na_2O$  and  $K_2O$ ), 5 to 35 percent boric oxides, and 0 to 13 percent other oxides.

Container glass means glass made of soda-lime recipe, clear or colored, which is pressed and/or blown into bottles, jars, ampoules, and other products listed in Standard Industrial Classification 3221 (SIC 3221).

Experimental furnace means a glass melting furnace with the sole purpose of operating to evaluate glass melting processes, technologies, or glass products. An experimental furnace does not produce glass that is sold (except for further research and development purposes) or that is used as a raw material for nonexperimental furnaces.

Flat glass means glass made of sodalime recipe and produced into continuous flat sheets and other products listed in SIC 3211.

Flow channels means appendages used for conditioning and distributing molten glass to forming apparatuses and are a permanently separate source of emissions such that no mixing of emissions occurs with emissions from the melter cooling system prior to their being vented to the atmosphere.

Glass melting furnace means a unit comprising a refractory vessel in which raw materials are charged, melted at high temperature, refined, and conditioned to produce molten glass. The unit includes foundations, superstructure and retaining walls, raw material charger systems, heat exchangers, melter cooling system, exhaust system, refractory brick work, fuel supply and electrical boosting equipment, integral control systems and instrumentation, and appendages for conditioning and distributing molten glass to forming apparatuses. The forming apparatuses, including the float bath used in flat glass manufacturing and flow channels in wool fiberglass and textile fiberglass manufacturing, are not considered part of the glass melting furnace.

Glass produced means the weight of the glass pulled from the glass melting furnace.

Hand glass melting furnace means a glass melting furnace where the molten glass is removed from the furnace by a glassworker using a blowpipe or a pontil.

Lead recipe means glass product composition of the following ranges of weight proportions: 50 to 60 percent silicon dioxide, 18 to 35 percent lead oxides, 5 to 20 percent total R<sub>2</sub>O (e.g., Na<sub>2</sub>O and K<sub>2</sub>O), 0 to 8 percent total R<sub>2</sub>O<sub>3</sub> (e.g., Al<sub>2</sub>O<sub>3</sub>), 0 to 15 percent total RO (e.g., CaO, MgO), other than lead oxide, and 5 to 10 percent other oxides.

Pressed and blown glass means glass which is pressed, blown, or both, including textile fiberglass, noncontinuous flat glass, noncontainer glass, and other products listed in SIC 3229. It is separated into:

- (1) Glass of borosilicate recipe.
- (2) Glass of soda-lime and lead recipes.